

**AMPHIDINOLIDE-A, A NOVEL ANTINEOPLASTIC MACROLIDE  
FROM THE MARINE DINOFLAGELLATE AMPHIDIINIUM SP.**

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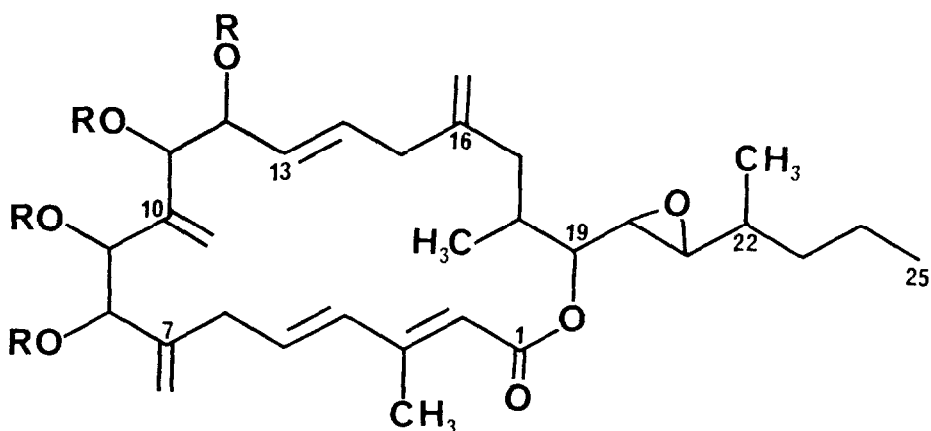
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**Summary:** A novel 20-membered macrolide, amphidinolide-A with antineoplastic activity has been isolated from the marine dinoflagellate Amphidinium sp. and its structure was determined to be 1 by spectral data.

Marine dinoflagellates have been demonstrated to produce chemically interesting and biologically significant metabolites, such as saxitoxin,<sup>1)</sup> brevetoxin,<sup>2)</sup> and okadaic acid.<sup>3)</sup> During our studies on bioactive substances from marine organisms,<sup>4)</sup> we have investigated extracts of some marine symbiotic microorganisms. In this communication, we describe the isolation and structure assignment of amphidinolide-A (1), a novel antineoplastic macrolide from a laboratory-cultured marine dinoflagellate Amphidinium sp.

The dinoflagellate was isolated from the Okinawan flatworm Amphiscolops sp. and grown unialgally in a sea water medium enriched with ES supplement<sup>5)</sup> at 25 °C for two weeks. The cultured cells were harvested by centrifugation to yield 375 g of cells from ca. 1000 L of culture. The extracts of harvested cells with methanol-toluene (3:1) were partitioned between toluene and water. The toluene soluble fraction was subjected to repeated silica-gel column chromatographies [methanol/chloroform (5:95) and hexane/acetone (2:1)] followed by reversed phase HPLC (ODS, 88% methanol) to give amphidinolide-A (1, 0.002% yield), which exhibited antitumor activity against L1210 (IC<sub>50</sub> 2.4 µg/mL) and L5178Y (IC<sub>50</sub> 3.9 µg/mL) murine leukemia cells in vitro, respectively.

Amphidinolide-A (1),<sup>6)</sup> colorless needles, mp 130-133 °C, [ $\alpha$ ]<sub>D</sub><sup>24</sup> +46° (c 1, CHCl<sub>3</sub>), was shown to have a molecular formula, C<sub>31</sub>H<sub>46</sub>O<sub>7</sub>, by high-resolution EI



1 R = H

2 R = Ac

Table 1.  $^1\text{H}$  (400 MHz) and  $^{13}\text{C}$  (100 MHz) NMR Spectra of Amphidinolide-A (1)  
( $\text{CDCl}_3$ ,  $\delta$  in ppm)

Position	H	$\text{C}^{\text{a,b}}$	Position	H	$\text{C}^{\text{a,b}}$
1		165.81 s	13	5.51 dd	130.96 d
2	5.80 s	118.62 d	14	5.69 dt	130.63 d
3		152.75 s	15	2.76 br s (2H)	39.78 t
Me-3	2.27 s	13.93 q	16		147.26 s
4	6.27 d	136.36 d	$\text{CH}_2=16$	4.79 s	112.79 t
5	6.09 ddd	134.72 d		4.88 s	
6	3.20 dd	38.96 t	17	2.35 dd	36.26 t
6'	3.13 dd		17'	1.92 dd	
7		144.88 s	18	2.17 m	33.37 d
$\text{CH}_2=7$	5.21 s	114.67 t	Me-18	1.06 d	15.81 q
	5.36 s		19	4.72 dd	70.62 d
8	4.42 br s	72.53 d	20	2.85 dd	54.27 d
9	4.58 br s	74.80 d	21	2.76 br s	61.85 d
10		145.05 s	22	1.38 m	35.36 d
$\text{CH}_2=10$	5.37 s	116.02 t	Me-22	0.93 d	14.93 q
	5.49 s		23	1.25-1.55 m (4H)	36.69 t
11	4.09 br s	73.51 d	24		19.96 t
12	4.22 br s	75.77 d	25	0.91 t (3H)	14.23 q

$J(\text{H,H})$  in Hz: 4,5=15.6; 5,6=4.8; 5,6'=8.8; 6,6'=14.6; 12,13=4.1; 13,14=15.4; 14,15=7.7; 14,15'=7.7; 17,17'=14.0; 17,18=5.5; 17',18=9.4; 18,Me(18)=7.0; 18,19=3.4; 19,20=5.9; 20,21=1.9; 22,Me(22)=6.0; 24,25(3H)=6.8.

a) Multiplicities were determined by DEPT data at 22.5 MHz.

b) Assignments were carried out on the basis of selective proton decoupling experiments at 22.5 MHz.

mass spectrum (Found  $m/z$  530.3210,  $\Delta$ -3.1 mmu). The UV absorption maximum at 265 nm ( $\epsilon$  18000, MeOH) indicated the presence of an  $\alpha,\beta,\gamma,\delta$ -dienoate chromophore.  $^{13}\text{C}$  NMR spectrum of **1** revealed the presence of the following groups: 4 x  $-\text{CH}_3$ , 5 x  $-\text{CH}_2-$ , 2 x  $-\overset{\cdot}{\text{C}}\text{H}-$ , 7 x  $-\overset{\cdot}{\text{C}}\text{H}-\text{O}$ , 3 x  $=\text{CH}_2$ , 5 x  $=\overset{\cdot}{\text{C}}\text{H}-$ , 4 x  $=\overset{\cdot}{\text{C}}-$ , and 1 x  $-\text{CO}_2-$ . This accounts for 31 carbons and 42 protons. The four missing protons belong to 4 hydroxyl groups which were confirmed by the acetylation of **1** to the tetraacetate [**2**,  $m/z$  698 ( $\text{M}^+$ )].<sup>7)</sup> Out of 7 methoxy carbon signals, four belong to hydroxy methine groups, one to a lactone terminal, and two remaining signals to the epoxide-bearing carbons, suggested by  $^1\text{H}$  and  $^{13}\text{C}$  signals ( $\delta_{\text{H}}$  2.85 dd and 2.76 br s;  $\delta_{\text{C}}$  61.85 d and 54.27 d).  $^{13}\text{C}$  NMR showed that **1** possesses six double bonds: three belong to *exo*-methylenes, two to a  $\alpha,\beta,\gamma,\delta$ -dienoate, and one to an isolated disubstituted double bond.

The detailed analyses of the COSY spectrum of **1** allowed the assignment of all protons (Table 1). The signal for H-2 showed two off-diagonal peaks, one due to allylic coupling to the methyl group on C-3 and the other allylic coupling to H-4. The H-4 signal was coupled to the vicinal proton on C-5 and two allylic protons on C-6. This observation revealed  $\beta$ -methyl- $\alpha,\gamma$ -diunsaturated ester moiety. The high-field shifted methyl carbon<sup>8)</sup> ( $\delta$ 13.93 q) on C-3 and the coupling constant of  $J_{4,5}$  (15.6 Hz) indicated  $2\text{E},4\text{E}$ -configuration.

The connectivities of the protons from  $\text{H}_2$  on C-6 to  $\text{H}_3$  on C-25 were clearly established by the COSY spectrum. The *exo*-methylene protons on C-7, C-10, and C-16 showed obvious cross peaks with the allylic protons on the neighboring carbons. The geometry of  $\Delta^{13,14}$ -double bond was determined to be *E* by the coupling constant ( $J_{13,14}$ =15.4 Hz). The signals for hydroxy methine protons (H-8, 9, 11, and 12) were shifted to lower field [ $\delta$ 5.35 (3H, m) and 5.93 (1H, s)] in the  $^1\text{H}$  NMR of the tetraacetate (**2**), while the signal for H-19 was observed at nearly the same chemical shift ( $\delta$ 4.72 for **1** and  $\delta$ 4.68 for **2**). This fact confirmed that the ester oxygen on C-1 is connected to C-19. The H-19 signal was vicinally coupled to the proton on epoxide-bearing carbon (H-20), which in turn was coupled to the other epoxide-methine proton (H-21) by 1.9 Hz, suggesting the epoxide to be *trans*.<sup>9)</sup> Finally, H-21 was found to be coupled to a methine proton (H-22) which was attached to the carbon with a secondary methyl group and a *n*-propyl group (C23-C25).

Amphidinolide-A (**1**) may be considered as a 20-membered macrolide with an alkyl side chain, bearing a sort of structural resemblance to swinholide A obtained from marine sponge *Theonella swinhoei*<sup>10)</sup> or scytophycin B from blue-green alga *Scytonema pseudohofmanni*.<sup>11)</sup> The compound **1** contains several interesting structural features. Three *exo*-methylene groups on the macrolide ring appear unique and four hydroxyl groups are located closely to each other (C8-C12) to form a hydrophilic part in the molecule.

Studies on the stereochemistry of **1** as well as other accompanying substances are under progress.

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#### References and Notes

- 1) Shimizu, Y. In "Marine Natural Products"; Scheuer, P. J. Ed.; Academic Press: New York, 1978: Vol. 1, pp. 1-42.
- 2) (a) Lin, Y. Y.; Risk, M.; Ray, S. M.; Van Enden, D.; Clardy, J.; Golik, J.; James, J. C.; Nakanishi, K. J. Am. Chem. Soc. **1981**, 103, 6773. (b) Shimizu, Y.; Chou, H.-N.; Bando, H.; Van Duyne, G.; Clardy, J. J. Am. Chem. Soc. **1986**, 108, 514.
- 3) (a) Tachibana, K.; Scheuer, P. J.; Tsukitani, Y.; Kikuchi, H.; Van Engen, D.; Clardy, J.; Gopichand, Y.; Schmitz, F. J. J. Am. Chem. Soc. **1981**, 103, 2469. (b) Murakami, M.; Oshima, Y.; Yasumoto, T. Bull. Jpn. Soc. Sci. Fish. **1982**, 48, 69. (c) Yasumoto, T.; Murata, M.; Oshima, Y.; Sano, M.; Matsumoto, G. K.; Clardy, J. Tetrahedron **1985**, 41, 1019.
- 4) (a) Kobayashi, J.; Nakamura, H.; Ohizumi, Y.; Hirata, Y. Tetrahedron Lett. **1986**, 27, 1191. (b) Nakamura, Y.; Kobayashi, J.; Gilmore, J.; Mascal, M.; Rinehart, K. L., Jr.; Nakamura, H.; Ohizumi, Y. J. Biol. Chem. **1986**, 261, 4139. (c) Kobayashi, J.; Ohizumi, Y.; Nakamura, H.; Hirata, Y. Tetrahedron Lett. **1986**, 27, 2113. (d) Kobayashi, J.; Ohizumi, Y.; Nakamura, H.; Hirata, Y.; Wakamatsu, K.; Miyazawa, T. Experientia in press.
- 5) Nakajima, I.; Oshima, Y.; Yasumoto, T. Bull. Jpn. Soc. Sci. Fish. **1981**, 47, 1029.
- 6) 1: IR (KBr) 3400, 1710, 1625, 1610, 1235, and 1160  $\text{cm}^{-1}$ ; CD (MeOH) 255 nm ( $\Delta\epsilon$  +7.0); MS (EI)  $m/z$  530 ( $M^+$ ), 512, and 494.
- 7) 2:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.90 (3H, t,  $J=7.0$  Hz;  $\text{H}_3$ -25), 0.92 (3H, d,  $J=6.6$  Hz; Me-22), 1.05 (3H, d,  $J=6.8$  Hz; Me-18), 1.99, 2.07, 2.10, and 2.24 (each 3H, s; Ac x 4), 2.33 (3H, s; Me-3), 3.14 (2H, m;  $\text{H}_2$ -6), 4.68 (1H, m; H-19), 4.78 and 4.83 (each 1H, s;  $\text{CH}_2=16$ ), 5.02 and 5.07 (each 1H, s;  $\text{CH}_2=7$ ), 5.20 (1H, m; H-13), 5.35 (3H, m), 5.36 and 5.48 (each 1H, s;  $\text{CH}_2=10$ ), 5.65 (1H, m; H-14), 5.84 (1H, s; H-2), 5.93 (1H, s), 6.17 (1H, m; H-5), and 6.58 (1H, d,  $J=16.2$  Hz; H-4); MS (EI)  $m/z$  698 ( $M^+$ ), 670, 638, 610, 578, and 307.
- 8) Carey, L.; Clough, J. M.; Pattenden, G. J. Chem. Soc. Perkin Trans I **1983**, 3005.
- 9) For 1,2-epoxypropane,  $J_1(\text{cis})_2=4$  Hz and  $J_1(\text{trans})_2=1.5$  Hz: In "Handbook of Proton NMR Spectra and Data"; Asahi Research Center, Ed.; Academic Press: Tokyo, 1985; Vol.1, p.86.
- 10) Carmely, S.; Kashman, Y. Tetrahedron Lett. **1985**, 26, 511.
- 11) Moore, R. E.; Patterson, G. M. L.; Mynderse, J. S.; Barch, J., Jr.; Norton, T. R.; Furusawa, E.; Furusawa, S. Pure Appl. Chem. **1986**, 56, 263.

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